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(54) Title: CURABLE COMPOSITIONS COMPOSITE COATINGS AND PROCESS FOR HAVING IMPROVED MAR AND ABRA- SION RESISTANCE			
(57) Abstract			
<p>A curable composition and process of applying the composition and the resulting composite coatings provide improved resistance to acid etch and mar. The curable composition has (a) about 30 to 90 percent by weight, based on total weight of resin solids, of a film-forming composition selected from the group consisting of: (i) a polyepoxide and a polyacid cross-linking agent; (ii) an acrylosilane polymer, an acrylic polyol polymer, and optionally an alkylated melamine-formaldehyde cross-linking agent; and (iii) a polyisocyanate and a polymer having at least one group that is reactive with isocyanate; and (b) about 10 to 70 percent by weight, based on total weight of resin solids, of a tricarbamoyl triazine compound of the formula: $C_3N_3(NHCOXR)_3$, wherein X is nitrogen, oxygen, sulfur, phosphorus, or carbon, and R is a lower alkyl group having 1 to 12 carbon atoms, or mixtures of lower alkyl groups. The process involves applying the curable coating as a monocoat on a cured primer surfacer coating on a substrate or as a clear coat on a curable base coat that are cured separately or conjointly. The composite coating includes the primer surfacer and monocoat on a substrate or the color-plus-clear coating on an electrodeposited coating layer or a primer surfacer coating layer.</p>			

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**CURABLE COMPOSITIONS COMPOSITE COATINGS AND
PROCESS FOR HAVING IMPROVED MAR AND ABRASION RESISTANCE**

5

BACKGROUND OF THE INVENTION

The present invention relates to curable compositions and composite coatings on substrates containing tricarbamoyl triazines and process for applying the composition for mar and abrasion resistance.

10 Of increasing concern in many geographic areas is the encounter with acidic precipitation. Coatings, particularly automotive original equipment coatings, that are resistant to such environmental conditions like those that are resistant to etching by atmospheric acid precipitation ("acid etch
15 resistance") are becoming increasingly desirable. Original equipment manufacturers are requiring that coating systems demonstrate acid etch resistance.

In addition to the requirement of acid etch resistance, original equipment coatings must also demonstrate mar and
20 abrasion resistance; i.e., the appearance of the coating must not be detrimentally affected when the coating is contacted with an abrasive material. Mar and abrasion resistance is a desirable property particularly due to the popularity of commercial car washes where brushes are often used to wash
25 car bodies.

A number of commercial coating systems which demonstrate acid etch resistance do not provide adequate mar and abrasion resistance. Microparticulate materials such as silica, metal sulfides, and crosslinked styrene-butadiene have been added
30 to these coating systems to improve mar resistance, but gloss and distinctness of image of the systems are adversely affected due to light scattering at the particle surfaces.

Recently, curing agents for coatings with active hydrogen-containing materials have been developed that are
35 triisocyanato or tricarbamoyl triazines as shown in U.S.

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Patent Nos. 4,930,213 and 5,084,541. The 4,930,213 ('213) patent shows the use of these curing agents in a coating composition to improve environmental resistance for powder coatings, coil coatings and can coatings. As noted on page 1
5 of the '213 patent, when the active hydrogen-containing compounds are hydroxylated polymers, environmentally-resistant coatings are provided.

It would be desirable to provide a curable composition for use as a coating which avoids the marring problems of the
10 prior art coating systems while maintaining excellent acid etch resistance and appearance properties.

SUMMARY OF THE INVENTION

In accordance with the present invention, a curable
15 composition, which is capable of exhibiting improved mar and acid etch resistance, is provided which comprises (a) about 30 to 90 percent by weight, based on total weight of resin solids, of a film-forming composition selected from the group consisting of:

20 (i) a polyepoxide and a polyacid crosslinking agent; (ii) an acrylosilane polymer, an acrylic polyol polymer, and optionally an alkylated melamine-formaldehyde crosslinking agent; and

25 (iii) a polyisocyanate and a polymer having at least one group that is reactive with isocyanate; and

(b) about 10 to 70 percent by weight, based on total weight of resin solids, of a tricarbamoyl triazine compound of the formula: $C_3N_3(NHCOXR)_3$, wherein X is nitrogen, oxygen, sulfur, phosphorus, or carbon, and R is a lower alkyl group
30 having 1 to 12 carbon atoms, or mixtures of lower alkyl groups.

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The process of the present invention and the composite coating involve applying a first coating as a primer surfacer coating layer to a substrate, curing the primer surfacer layer, and applying a second or an additional coating layer
5 which is the aforedescribed tricarbamoyl triazine compound containing curable coating composition, and curing this coating.

The invention also provides a process for applying a color-plus-clear composite coating to a substrate comprising
10 applying to the substrate a pigmented or colored first curable coating composition to form a base coat and applying to the base coat a second curable coating composition which is transparent to form a clear coat over the base coat. The clear coat can be the curable composition of the present
15 invention.

DETAILED DESCRIPTION

In the following description, reference to number average and weight average molecular weights refers to
20 measurements determined by gel permeation chromatography using a polystyrene standard as is well known to those skilled in the art. Where ranges of amounts are stated in percentages for components that are combined to yield a composition, the amount of each of those various components
25 yielding the composition are generally selected from each range for that component so that the total amount of the components in the composition equals around 100 percent.

In the curable composition of the present invention, the aforedescribed film-forming compositions each is capable of
30 film formation with crosslinking. Crosslinking can be through the one or more designated polymeric materials with the crosslinking material. For the film-forming compositions

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with the additional polymeric material or with two polymeric materials that are reactive with each other, the additional polymeric material and/or the inter-reactivity of both polymers results in crosslinking. The polyepoxide and a 5 polyacid film-forming composition has the polyacid that is used as the crosslinking agent since the polyepoxide is the major component of the film. For the film-forming composition of the acrylosilane polymer and the acrylic polyol polymer, these polymeric materials have functionality 10 so that one reacts to form crosslinks with the other along with any alkylated melamine-formaldehyde crosslinking agent. This occurs in a similar manner in the film-forming composition of polyisocyanate and a polymer having at least one group that is reactive with isocyanate. It is believed 15 without limiting the present invention that the tricarbamoyl triazine compound is a crosslinking agent that also provides crosslinking in addition or further to that of the aforementioned film-forming compositions.

The film-forming composition in the curable composition 20 of the present invention preferably comprises a mixture of a polyepoxide and a polyacid crosslinking agent. Preferably, the polyepoxide is present in the film-forming composition (i) in amounts of about 10 to 90, more preferably from about 25 to 50 percent by weight based on total weight of resin 25 solids in the film-forming composition (i). Preferably, the polyepoxides have a glass transition temperature (Tg) less than 50°C, more preferably less than 30°C. The Tg is described in *PRINCIPLES OF POLYMER CHEMISTRY*, Flory, Cornell University Press, Ithaca, New York, 1953, pages 52-57. The 30 Tg can be calculated as described by Fox in *Bull. Amer. Physic. Society*, 1,3, page 123 (1956). The Tg can be measured experimentally by using a penetrometer such as a

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DuPont 940 Thermomedian Analyzer. The Tg of the polymers as used herein refers to the calculated values unless otherwise indicated.

Among the polyepoxides which can be used are epoxy-
5 containing acrylic polymers which are preferred, epoxy condensation polymers such as polyglycidyl ethers of alcohols and certain polyepoxide monomers and oligomers. Epoxy- containing acrylic polymers are preferred because they yield products which have the optimum combination of coating
10 properties; i.e., smoothness, gloss, durability, and solvent resistance. The epoxy-containing acrylic polymer is a copolymer of an ethylenically unsaturated monomer having at least one epoxy group and at least one polymerizable ethylenically unsaturated monomer which is free of epoxy
15 groups. The preparation of the epoxy-containing acrylic polymer may be conducted as disclosed in U.S. Patent No. 4,650,718, incorporated herein by reference. The epoxy- containing acrylic polymer typically has a number average molecular weight between about 1000 and 20,000, preferably
20 about 1000 to 10,000, and more preferably about 1000 to 5000.

The ethylenically unsaturated monomers useful in preparing the epoxy-containing acrylic polymer can be selected from numerous art-recognized monomers such as the below-described nonexclusive examples. Examples of
25 ethylenically unsaturated monomers containing epoxy groups are those containing 1,2-epoxy groups and include glycidyl acrylate, glycidyl methacrylate, and allyl glycidyl ether. Examples of ethylenically unsaturated monomers which do not contain epoxy groups are vinyl monomers and alkyl esters of
30 acrylic and methacrylic acid containing from 1 to 20 atoms in the alkyl group such as those art-recognized monomers

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disclosed in U.S. Patent No. 4,650,718, incorporated herein by reference.

The epoxy group-containing ethylenically unsaturated monomer is preferably used in amounts of from about 10 to 65, 5 more preferably from 20 to 50 percent by weight of the total monomers used in preparing the epoxy-containing acrylic polymer. Of the remaining polymerizable ethylenically unsaturated monomers, preferably from 35 to 90 percent, more preferably from 50 to 80 percent by weight of the total 10 monomers are the alkyl esters of acrylic and methacrylic acid containing from 1 to 20 carbon atoms in the alkyl group, and about 0 to 50 percent, preferably from about 5 to 40 percent based on weight of total monomers are other ethylenically unsaturated monomers; i.e., the vinyl aromatic compounds, 15 nitriles, vinyl and vinylidene halides and vinyl esters.

The epoxy condensation polymers which may be used are polyepoxides; that is, those having a 1,2-epoxy equivalency greater than 1, preferably greater than about 1 and up to about 3. The preferred polyepoxides are polyglycidyl ethers 20 of aliphatic alcohols. These polyepoxides can be produced by etherification of alcohols, such as those known in the art and described in U.S. Patent No. 4,650,718, with an epihalohydrin or dihalohydrin such as epichlorohydrin or dichlorohydrin in the presence of alkali.

25 In addition to the epoxy-containing acrylic polymers and epoxy condensation polymers described above, certain polyepoxide monomers and oligomers can also be used. Examples of these materials are described in U.S. Patent No. 4,102,942 in column 3, lines 1-16, incorporated herein by 30 reference. Specific examples of such low molecular weight polyepoxides are 3,4-epoxycyclohexylmethyl 3,4-epoxycyclohexane-carboxylate and bis(3,4-

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epoxycyclohexylmethyl) adipate. These low molecular weight polyepoxides may be used to increase the cure response and solids content of the curable compositions. When used, they are present in amounts up to 30 percent by weight based on
5 the total weight of resin solids in the crosslinkable composition.

The film-forming composition (i) in the curable composition of the present invention further includes a polyacid crosslinking agent present in the film-forming
10 composition in amounts of about 10 to 90, preferably 25 to 75 percent by weight based on total weight of resin solids in the film-forming composition. The polyacid crosslinking agent has a high average acid functionality. More specifically, the polyacid crosslinking agent on average
15 contains more than two acid groups per molecule, more preferably three or more, and most preferably four or more, such acid groups being reactive with the polyepoxide to form a crosslinked coating as indicated by its resistance to organic solvent. The parameter of greater than two acid
20 groups per molecule is intended to encompass mixtures of polyacid crosslinking agents in which di-functional curing agents are mixed with tri- or higher functionality polyacid crosslinking agents. Polyacid crosslinking agent mixtures including up to about 50 percent of a di-functional curing
25 agent with a tri-functional curing agent are suitable. Higher percentages of di-functional material can be used if the remainder of the curing agent mixture is higher than tri-functional or if the polyacid crosslinking agent mixture is used with a highly functional polyepoxide component. The
30 acid functionality is preferably carboxylic acid, although acids such as phosphorus-based acid may be used. Preferably, the polyacid crosslinking agent is a carboxylic acid

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terminated material having, on average, greater than two carboxylic acid groups per molecule. Among the polyacid crosslinking agents which may be used are carboxylic acid group-containing polymers such as acrylic polymers,

5 polyesters, and polyurethanes; oligomers such as ester group-containing oligomers, which are preferred; and monomers. Suitable art-recognized polyacid crosslinking agents include those described in U.S. Patent Nos. 4,650,718; 4,681,811; and 4,703,101, incorporated herein by reference.

10 The equivalent ratio of the reactants present in the film-forming composition (i) is adjusted such that for each equivalent of epoxy there are around 0.3 to around 3.0, preferably 0.75 to 1.5 equivalents of carboxyl (anhydride, if present, is considered monofunctional).

15 The film-forming composition (i) may also contain additional materials as known to those skilled in the art. For example, an anhydride for improved cure response and copolymer of an alpha olefin and olefinically unsaturated anhydride for improved humidity resistance of the cured 20 coating can be used. The anhydrides are preferably liquids at 25°C. Examples of suitable anhydrides include dodecenyl succinic anhydride and alkyl-substituted hexahydrophthalic anhydrides wherein the alkyl group contains up to 7 carbon atoms, more preferably up to 4 carbon atoms, such as methyl 25 hexahydrophthalic anhydride. The amount of the anhydride which is used in the curable composition can vary from about 0 to 40 percent, preferably from about 5 to 25 percent by weight based on total weight of resin solids of the film-forming composition. An example of the copolymer is one of 30 1-octene or 1-decene and maleic anhydride. Further art-recognized materials and their amounts are described more

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fully in U.S. Patent No. 4,927,868, incorporated herein by reference.

To form one-package compositions, the film-forming composition (i) in the curable composition of the present invention is substantially free of basic esterification catalyst. The high functionality associated with the polyepoxide and polyacid provide for sufficient cure response. Although the absence of catalyst may have a negative effect on the cure of the composition, it provides for a stable composition and is also beneficial in reducing or eliminating cure inhibition between layers in a color-plus-clear formulation when the base coat contains an acid-catalyzed resinous binder. As is known in the art, properties of multilayered composite coatings can be deleteriously affected by different cure chemistries between coating layers which interfere with each other; e.g., an acid-catalyzed base coat and a base-catalyzed clear coat. In a preferred embodiment, the film-forming composition (i) of the present invention has no or only small amounts of basic esterification catalyst such that the composition is stable for a time sufficient to allow formulation of the composition as a single component; i.e., one-package composition.

To form multi-package or multi-component compositions in which the polyepoxide and polyacid crosslinking agent are present in separate packages and combined shortly before application, an esterification catalyst to promote cure can be included in the film-forming composition (i). A number of such catalysts are known in the art. These catalysts include basic materials such as secondary amine catalysts, for example, piperidine; tertiary amine catalysts such as N,N-dimethyldodecylamine, pyridine, and N,N-dimethylaniline; ammonium compounds, including tetrabutylammonium bromide,

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tetrabutylammonium hydroxide, and tetrabutylammonium acetate; phosphonium compounds, including ethyltriphenylphosphonium acetate and tetrabutyl phosphonium bromide; and other ammonium and phosphonium salts.

5 In another embodiment of the invention, the film-forming composition (ii) in the curable composition of the present invention comprises an acrylosilane polymer, an acrylic polyol polymer, and optionally an alkylated melamine-formaldehyde crosslinking agent.

10 The acrylosilane polymer is prepared by polymerizing about 40 to 95 percent by weight ethylenically unsaturated non-silane containing monomers, and about 5 to 60 percent by weight ethylenically unsaturated alkoxy or acyloxy silane monomers, based on the weight of the acrylosilane. Suitable

15 ethylenically unsaturated non-silane containing monomers include alkyl and aryl acrylates and methacrylates having 1 to 12 carbon atoms in the alkyl or aryl groups as are known to those skilled in the art. The alkyl groups also may be cycloaliphatic. Other suitable ethylenically unsaturated

20 non-silane containing monomers include styrene, methyl styrene, acrylamide, acrylonitrile, and the like, and non-interfering mixtures thereof; i.e., mixtures which do not interfere with any crosslinking mechanisms of the film-forming composition. Suitable ethylenically unsaturated

25 alkoxy or acyloxy silane monomers include acrylatoalkoxy silanes, methacrylatoalkoxy silanes vinylalkoxy silanes, and acyloxy silanes such as acrylatoxy silane, methacrylatoxy silane, and vinyl-acetoxy silanes, and non-interfering mixtures thereof.

30 The acrylosilane polymer typically has a weight average molecular weight of about 1,000 to 30,000 and a number average molecular weight of about 500 to 6,000. One or more

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of these types of acrylosilane polymers are present in the film-forming composition (ii) in amounts of about 10 to 90 percent by weight based on total weight of resin solids in the film-forming composition (ii).

- 5 The film-forming composition (ii) further includes an acrylic polyol polymer present in the film-forming composition in amounts of about 10 to 90 percent by weight based on total weight of resin solids in the film-forming composition (ii). The acrylic polyol polymer preferably has
10 a hydroxyl number of about 50 to 200, and a weight average molecular weight of about 1,000 to 200,000, preferably about 1,000 to 20,000. The hydroxyl number can be determined by any art-recognized method such as by theoretical calculation or by analytical methods.
- 15 The acrylic polyol polymer is prepared by polymerizing one or more hydroxyl functional alkyl acrylate or methacrylate monomers and up to about 80 percent by weight ethylenically unsaturated non-hydroxyl functional monomers, based on the weight of the acrylic polyol. Suitable hydroxyl
20 functional alkyl acrylate or methacrylate monomers include hydroxy ethyl acrylate and methacrylate, hydroxypropyl acrylate and methacrylate, hydroxybutyl acrylate and methacrylate, and the like. Suitable ethylenically unsaturated non-hydroxyl functional monomers include alkyl
25 and aryl acrylates and methacrylates having 1 to 16 carbon atoms in the alkyl or aryl groups as are known to those skilled in the art. Other suitable ethylenically unsaturated non-hydroxyl functional monomers include styrene, methyl styrene, acrylamide, acrylonitrile, and the like.
- 30 The film-forming composition (ii) containing the acrylosilane polymer and acrylic polyol polymer may further include an alkylated melamine-formaldehyde crosslinking agent

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present in the film-forming composition in amounts of up to about 50 percent by weight based on total weight of resin solids in the film-forming composition (ii).

The melamine-formaldehyde crosslinking agent may be 5 monomeric or polymeric and may be partially or fully alkylated. This crosslinking agent typically has a weight average molecular weight of about 500 to 1500, and a number average molecular weight of about 300 to 600. Suitable crosslinking agents include CYMEL 1168, CYMEL 1161, and CYMEL 10 1158, all available from CYTEC Industries, Inc., Five Garret Mountain Plaza, West Paterson, New Jersey 07424, and RESIMENE 755 and RESIMENE 4514, available from Monsanto Chemical Co.

In another embodiment of the invention, the film-forming composition (iii) in the curable composition of the present 15 invention is a composition which may be single- or multi-package comprising a polyisocyanate or isocyanate functional resin or polymer and a polymer having at least one group that is reactive with isocyanate. The film-forming composition may further include active hydrogen containing reactive 20 diluents as are known to those skilled in the art.

Polyisocyanate may be aliphatic, cycloaliphatic, or heterocyclic and may be unsubstituted or substituted. These can range from the simple isocyanates such as isophorone diisocyanate, trimethylhexamethylene diisocyanate, 25 dicyclohexylmethane diisocyanate to more elaborate isocyanurates. Examples of the latter include the isocyanurate available from Mobay Chemical, now Miles or Bayer U.S.A., as "Des N-3390"; isocyanurate derived from isophorone diisocyanate, commercially available from VEBA 30 Company as T1890. Also, biurets such as the biuret from 1,6-hexamethylene diisocyanate, commercially available from Mobay Chemical, now Miles or Bayer U.S.A., as DESMODUR N,

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mixtures of polyisocyanates and monoisocyanates, isocyanate prepolymers, and mixtures of organic isocyanate prepolymers with monomeric isocyanates (so-called semi-prepolymers) may be utilized as the polyisocyanate. The isocyanate
5 prepolymers can be prepared by any method known to those skilled in the art such as those along with further examples of the polyisocyanates as shown in U.S. Patent Nos. 4,798,746 (Claar, et al.); 4,798,745 (Martz, et al.); 4,699,814 (Ambrose, et al.); 3,567,692 (Haggis et al.) and Canadian
10 patent document 1262596, all hereby incorporated by reference.

The polyisocyanate is preferably a fully-capped polyisocyanate with substantially no free isocyanate groups. The polyisocyanate is preferably aliphatic. Diisocyanates
15 are preferred, although higher polyisocyanates can be used in place of or in combination with diisocyanates. Examples of suitable aliphatic diisocyanates are straight chain aliphatic diisocyanates such as 1,4-tetramethylene diisocyanate and 1,6-hexamethylene diisocyanate, and isocyanurates thereof.
20 Also, cycloaliphatic diisocyanates can be employed. Examples include isophorone diisocyanate, isocyanurates of isophorone diisocyanate, and 4,4'-methylene-bis-(cyclohexyl isocyanate), and Metaxylylene diisocyanate. Suitable capping agents include those well known in the art such as pyrazoles as
25 those described in U.S. Patent No. 5,352,755, alcohols, lactams, oximes (which are preferred), malonic esters, phenols, amines, and the like.

The polyisocyanate is present in the film-forming composition (iii) in amounts of about 10 to 60 percent by weight, preferably 15 to 40 percent by weight based on total
30 weight of resin solids in the film-forming composition (iii).

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Suitable polymers having at least one group that is reactive with isocyanate include those having hydroxyl groups, amino groups, thiol groups, hydrazide groups, and the like. Preferably, the reactive group is hydroxyl. The 5 polymer may be an acrylic, polyester, polyepoxide, polycarbonate, polyurethane, polyamide, polyimide, polysiloxane, acrylic polyols; polyester polyols; polyether polyols; amide-containing polyols; epoxy polyols; polyhydric polyvinyl alcohols; cellulose and derivatives thereof, 10 urethane polyols; and mixtures thereof.

The polymer is preferably an acrylic. The polymer is present in the film-forming composition in amounts of about 40 to 90 percent by weight, preferably 60 to 85 percent by weight based on total weight of resin solids in the clear 15 coat composition.

For instance, one such suitable film-forming composition for (iii) is a reaction product of uncapped or partially or completely capped, preferably uncapped, polyisocyanates which contain isocyanurate groups and may or may not contain biuret 20 groups, have a functionality of from 2.5 to 6, preferably from 3 to 5, and possess predominantly cycloaliphatically bonded isocyanate groups, with, as a polyol mixture, hydroxyl-containing polyacrylates of specific composition and polyesterols which are soluble in organic solvents. In this 25 case the reaction product can be of a first mixture of (1) a polyhydroxypolyacrylate resin obtained from esters of acrylic acid and/or methacrylic acid with monofunctional aliphatic alcohols, monoesters of acrylic acid and/or methacrylic acid with polyfunctional aliphatic alcohols and other 30 copolymerizable olefinically unsaturated monomers, and (2) a polyesterpolyol which is soluble in organic solvents with a polyisocyanate which is uncapped or partially or completely

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capped and contains isocyanurate groups and may or may not contain biuret groups and has a functionality of from 2.5 to 6. Other formulations known to those skilled in the art for film-forming composition (iii) can also be used, and these 5 formulations can be those referred to as two-pack isocyanate coating formulations.

The tricarbamoyl triazine compound in the curable composition of the present invention has the formula $C_3N_3(NHCOXR)_3$, wherein X is nitrogen, oxygen, sulfur, 10 phosphorus, or carbon, and R is a lower alkyl group having 1 to 12 carbon atoms, or mixtures of lower alkyl groups. The X moiety is preferably oxygen or carbon, more preferably oxygen. The R moiety preferably has one to eight carbon atoms, for example, methyl, ethyl, n-propyl, i-propyl, butyl, 15 n-octyl, 2-ethylhexyl, and R is preferably a mixture of methyl and butyl groups. Such compounds and the preparation thereof are described in detail in U.S. Patent No. 5,084,541, incorporated herein by reference. For convenience, these compounds in the remaining portion of this disclosure may be referred to as "TACT". These compounds are not aminoplasts, 20 which are condensates obtained from the reaction of an aldehyde such as formaldehyde, such as acetaldehyde, crotonaldehyde, and benzaldehyde with an amine or amide such as melamine, urea, or benzoguanamine.

25 The tricarbamoyl triazine compound in the curable composition of the present invention is present in amounts of about 10 percent to 70 percent by weight, more preferably from about 10 to 40 percent by weight, based on total weight of resin solids in the curable composition. It may be added 30 to the film-forming composition neat, that is, added by itself or in other resinous ingredients, or with solvents or other diluents. In the case of a multi-pack film-forming

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composition, the tricarbamoyl triazine may be mixed with either component before application of the composition to a substrate.

Though not intending to be bound by theory, it is
5 believed that the tricarbamoyl triazine compound in the
curable composition of the present invention reacts with
active hydrogens such as hydroxyl, carboxyl, amido, primary
and secondary amino, thiol, and the like, present in or
generated during cure in the film-forming composition,
10 thereby "co-curing" or post-curing in the composition.

Other optional ingredients, such as plasticizers,
catalysts, anti-oxidants, hindered amine light stabilizers,
UV light absorbers and stabilizers may be formulated into the
curable compositions of the present invention. When used in
15 a manner as is known to those skilled in the art, one can
appreciate that these ingredients are present (on an
individual basis) in amounts up to 10 percent, preferably
from about 0.1 to 5 percent by weight based on total weight
of resin solids of the curable composition. Organic and
20 inorganic pigments, including metallic pigments such as those
disclosed below with respect to base coats may also be
formulated into the curable compositions of the present
invention in art-recognized amounts.

The curable compositions preferably are formulated into
25 liquid high solids coating compositions, that is,
compositions containing greater than 40 percent, preferably
greater than 50 percent by weight resin solids. The solids
content is determined by heating a sample of the composition
to 105-110°C for 1-2 hours to drive off the volatile material
30 and measuring relative weight loss. Although the
compositions are preferably liquid coating compositions, they
may be formulated as powder coating compositions.

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The curable compositions of the invention may be applied to a substrate by any conventional coating technique such as brushing, spraying, dipping or flowing, but spray applications are preferred because of superior gloss. Any of 5 the known spraying techniques may be employed such as compressed air spraying, electrostatic spraying and either manual or automatic methods.

After application of the coating composition to the substrate, the coated substrate is heated to cure the 10 coating. In the curing operation, solvents are driven off and the film-forming materials of the coating composition are crosslinked. The heating or curing operation is usually carried out at a temperature in the range of from 160-350°F (71-177°C) but if needed lower or higher temperatures may be 15 used as necessary to activate crosslinking mechanisms. The thickness of the coating is usually from about 0.5-5 mils (12.7-127 microns), preferably 1.2-3 mils (30.5-76.2 microns).

Preferably, the compositions of the present invention 20 are used to formulate clear coats for use in a color-plus-clear application. In a color-plus-clear application, a composite coating is applied to a substrate. The process comprises applying to the substrate a pigmented or colored film-forming composition to form a base coat and applying to 25 the base coat a second film-forming composition to form a transparent clear coat, usually the top coat over the base coat.

The clear film-forming composition which forms the transparent coat which is usually the top coat is a curable 30 coating composition comprising the film-forming composition of (i) or (ii) of (iii) and the TACT material. The application of the clear coat may vary slightly depending on

the type of film-forming composition that is used. Generally, the process includes applying to the substrate the pigmented or colored film-forming composition to form a base coat and applying to the base coat curable coating

5 composition to form a transparent top coat, or clear coat, over the base coat. The clear top coat composition is applied to the base coat by any of the conventional coating techniques mentioned above, with spray applications preferred. Typically, the clear coat is applied to the base
10 10 coat via a wet-on-wet technique before the base coat has been cured. The two coatings are then heated to conjointly cure both coating layers. Curing conditions such as described above may be used.

In a particularly preferred embodiment, additional
15 15 coating layers such as a primer and/or a primer-surfacer may be applied to the substrate prior to application of the composition of the present invention, or when the composition of the present invention is used as a clear coat in a color-plus-clear application, additional coating layers such as a
20 20 primer and/or a primer-surfacer may be applied to the substrate prior to application of the base coat.

Suitable primers and primer-surfacers include electrodeposable and spray applied primers, as are known to those skilled in the art. Examples of suitable primers
25 25 include several available from PPG Industries, Inc., as DPX-1791, DPX-1804, DSPX-1537, and GPXH-5379.

The "primer surfacer", which is a primer composition for use under a subsequently applied topcoating composition includes such materials as thermoplastic and/or crosslinking
30 30 (e.g., thermosetting) film-forming resins generally known in the art of organic coating compositions. The term "thermosetting" is intended to include those that are capable

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of crosslinking either with or without the application of heat. These include thermoplastic resins and/or thermosetting resins generally known as cellulosics, acrylics, aminoplasts, urethanes, polyesters, epoxies and 5 polyamides, all as further described in U.S. Patent No. 5,066,733 (Martz, et al.). These resins, when desired, may also contain functional groups characteristic of more than one class, as for example, polyester amides, uralkyds, urethane acrylates, urethane amide acrylates, and the like.

10 As is described in U.S. Patent No. 5,356,973, incorporated herein, the spray applied primer surfacer can be applied to the electrocoat before topcoating. For example, substrates such as panels can be electrocoated with ED-11 electrodepositable coating from PPG Industries Inc. and can 15 be primed with a commercially available PPG European primer surfacer coded E 730G305. This primer surfacer can be cured for 25 minutes at 329°F (165°C). Another example of a suitable primer surfacer can be the two-package, acrylic urethane primer surfacer known as K200/K201 more fully disclosed in U.S. Patent No. 5,239,012 (McEntire, et al.) and 20 U.S. Patent No. 5,182,355, incorporated herein for such primer surfacer. This primer is preferably sanded with No. 400 grit paper and sealed with DP-40/401, which is a two-component epoxy primer which was reduced 100 percent by 25 volume with a thinner, DTU 800. The K200/K201, DP-40/401 and DTU 800, all as available from PPG Industries, Inc.

An additional primer-surfacer is that available from PPG Industries, Inc. as E-5584. It is reducible with 2-butoxyethylacetate to a viscosity of 23 seconds as measured 30 with a No. 4 Ford cup. This primer surfacer can be sprayed automatically and cured by flashing at ambient conditions for 15 minutes followed by heating for around 30 minutes at

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around 325°F (165°C) to produce coatings that can have dry film thickness of around 30 microns. The cured film is preferably sanded smooth with 500 grit sandpaper. Useful automatic spraying for both the primer surfacer and the clear 5 coat is the SPRAYMATION 310160 Automatic Test Panel Spray Unit available from SPRAYMATION Inc. The useable spray gun is a Binks Model 610, with open gun pressure 60 psi (4.22 kg/cm²) and traverse speed of around 80 rpm.

Another suitable primer surfacer can be the water 10 dispersed primer-surfacer composition with a combination of three essential resinous film-forming or binder ingredients as disclosed in U.S. Patent No. 4,303,581 (Levine, et al.), herein incorporated. The primer surfacer has a) 50 to 90 percent of a high molecular addition copolymer of a styrenic 15 monomer with acrylic monomers in latex form, b) about 5 to 40 percent of a water soluble epoxy ester resin, and c) about 5 to 20 percent of at least one water soluble or water dispersible aminoplast resin. All percents are based on percent by weight of the total of the binder ingredients.

With the use of an electrocoating composition such as 20 that of U.S. Patent No. 4,615,779 (McCollum, et al.), hereby incorporated, which is capable of providing a smooth coating on substrates, the presence of an intermediate coating primer surfacer (often sanded) can be avoided. Such an electrocoat 25 is available commercially as UNIPRIME® from PPG Industries, Inc.

The film-forming composition of the base coat can be any 30 of the compositions useful in coatings applications, particularly automotive applications. The film-forming composition comprises a resinous binder and a pigment to act as the colorant. Particularly useful resinous binders are acrylic polymers, polyesters, including alkyds, and

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polyurethanes, which are described in detail in U.S. Patent Nos. 4,650,718 and U.S. 5,066,733 (Martz, et. al) as the second organic coating, both of which are incorporated herein by reference.

5 The base coats have polymers as resinous binders that can be organic solvent-based polymers, although acrylic polymers can be prepared via aqueous emulsion polymerization techniques and used as aqueous-based base coat compositions. Also, water-based base coats in color-plus-clear compositions
10 are disclosed in U.S. Patent No. 4,403,003, incorporated herein by reference, and the resinous compositions used in preparing these base coats can be used in the practice of this invention. Also, water-based polyurethanes such as those prepared in accordance with U.S. Patent No. 4,147,679,
15 incorporated herein by reference, can be used as the resinous binder in the base coat. Further, water-based coatings such as those described in U.S. Patent No. 5,071,904, incorporated herein by reference, can be used as the base coat.

The base coat also contains pigments and other
20 additional components traditionally used in base coat formulations. Useful pigments to give color to the base coat include metallic or non-metallic pigments. The use of metallic flake pigmentation yield the so-called "glamour metallic" finishes, intended most often for the surface of
25 automobile bodies. Further nonexclusive examples of the pigments and other additives including their types, amounts, and proper use that are useful in the base coat are given in U.S. Patent Nos. 5,401,790 (Poole, et al.); 5,384,367 (Swarup, et al.); and 5,256,452. It is preferred that the
30 base coat have at least one or more pigments, where metallic pigments can be aluminum flake, copper bronze flake or mica, and non-metallic pigment can be inorganic pigments such as

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titanium dioxide, iron oxide, chromium oxide, lead chromate, and carbon black, or organic pigments such as phthalocyanine blue and phthalocyanine green. In general, the pigment is incorporated into the coating composition in amounts of about 5 1 to 80 percent by weight based on weight of coating solids, where up to 60 percent gives better results. The metallic pigment is employed in amounts of about 0.5 to 25 percent by weight based on weight of coating solids.

As mentioned in the afore-incorporated patents, the 10 various additional components that are well known in the art can include surfactants, UV stabilizers, flow control agents, thixotropic agents, fillers, anti-gassing agents, organic cosolvents, catalysts, and other customary auxiliaries. These materials can constitute up to 40 percent by weight of 15 the total weight of the coating composition.

The base coating compositions can be applied to various substrates to which they adhere. The compositions can be applied by conventional means including brushing, dipping, flow coating, spraying and the like, but they are most often 20 applied by spraying. The usual spray techniques and equipment for air spraying and electrostatic spraying by either manual or automatic methods can be used. Examples of these substrates are wood, metals, glass, cloth plastic, foam, including elastomeric substrates and the like. The 25 base coats are particularly useful in applications over metals and elastomeric substrates that are found on motor vehicles.

During application of the base coat composition to the substrate, a film of the base coat is formed on the 30 substrate. Typically, the base coat thickness will be about 0.01 to 5, preferably 0.1 to 2 mils in thickness. The film is formed after application by driving solvent, i.e., organic

solvent or water, out of the base coat film by heating or by an air drying period. Preferably, the heating will only be sufficient for a short period of time to ensure that the clearcoat can be applied to the base coat without the former 5 dissolving the base coat composition. Suitable drying conditions will depend on the particular base coat composition and on the ambient humidity with certain water-based compositions, but in general a drying time of from about 1 to 5 minutes at a temperature of about 80-250°F (20-10 121°C) will be adequate to ensure that mixing of the two coats is minimized. At the same time, the base coat film is adequately wetted by the topcoat composition so that satisfactory intercoat adhesion is obtained. Also, more than one base coat and multiple clear coats may be applied to 15 develop the optimum appearance. Usually between coats, the previously applied coat is flashed that is, exposed to ambient conditions for about 1 to 20 minutes.

The curable composition of the present invention as the clear coat composition, which is preferably the topcoat, is 20 applied to the base coat by any of the conventional coating techniques mentioned above, with spray applications preferred. Typically, the clear top coat is applied to the base coat via a wet-on-wet technique before the base coat has been cured. The two coatings are then heated to conjointly 25 cure both coating layers. Curing conditions such as described above may be used. The preferred curable composition of the present invention is where the film-forming composition is (i) the polyepoxide and a polyacid crosslinking agent used with the TACT in the aforementioned 30 preferred amounts.

The curable compositions of the present invention may be used as monocoats or as a component of a composite coating

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and have excellent etch resistance properties. The monocoat is any topcoat which can be pigmented or unpigmented that is on top of a previous primer type of coating. As used herein, the term "etch resistance" refers to the ability of a cured 5 composition to resist etching by acids and water spotting. Etch resistance is typically evaluated by visual inspection of coated substrates after actual or simulated weathering. It should be noted that simulated weathering typically, but not always, corresponds to actual weathering. Moreover, it 10 should be noted that cured compositions may have different etch resistance properties when subjected to actual weathering in different geographic locations. An etch resistant composition, as discussed herein, refers to a composition which has etch resistant properties under actual 15 weathering in at least one geographic site or which has etch resistant properties under simulated weathering. In addition to the acid etch properties, the curable compositions of the present invention also can have good mar or abrasion resistance properties, evaluated by measuring the gloss of 20 coated substrates before and after abrading of the coated substrates using a consistent laboratory method. To improve total performance even further, the primer surfacer coating can be used in a composite coating with the curable composition of the present invention, alone or in a color- 25 plus-clear coating composite, on the substrate.

The invention will further be described by reference to the following examples. Unless otherwise indicated, all parts are by weight.

30

EXAMPLES 1 THROUGH 4

Examples 1 through 4 illustrate the preparation of various coating compositions in accordance with the present

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invention which are used in a color-plus-clear composite coating process. Example 1 is a control two-component clear coat containing polyisocyanates and an acrylic polyol.

Example 2 illustrates the preparation of a clear coat

5 containing a tricarbamoyl triazine compound in place of one of the polyisocyanates. Example 3 is generally the same as Example 2 with the tricarbamoyl triazine compound in place of the other polyisocyanate, and Example 4 illustrates the preparation of a clear coat containing a tricarbamoyl

10 triazine compound in place of one half of the solid amount of a polyisocyanate.

The coating compositions of Examples 1 through 4 were prepared by mixing the ingredients together for each component in the order listed. A pigmented base coat,

15 available from PPG Industries, Inc. as BWB8555, was spray applied in two coats with a 90 second flash between coats to electrocoated steel panels at room temperature (67°F, 19.4°C) and at a relative humidity between 30 and 35 percent, to achieve a dry film thickness of 0.6 mil (15.2 microns).

20 After applying the second coat, a 5 minute flash at 200°F (93.3°C) was allowed before application of the coating compositions of Examples 1 through 4. The respective components for each example were combined and the compositions were applied in two coats with a 90 second room

25 temperature flash between coats, to achieve a dry film thickness of about 2 mil (50.8 microns). Each panel was given a 15 minute flash at room temperature and then cured for 30 minutes at 250°F (121.1°C).

Mar resistance of coated panels was measured using the

30 following method: Gloss of coated panels is measured with a 20° BYK Gardner GLOSSGARD II Glossmeter, available from Gardner Instrument Co. Coated panels are marred by applying

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a dry abrasive powder cleanser (Bon Ami™ cleanser, Faultless Starch/Bon Ami Co.) followed by ten double rubs to the surface with a wool felt cloth using a Crockmeter mar tester (available from Atlas Electric Devices Company). The 20
 5 degree gloss is read on the marred area of the panel after being washed with water and patted dry. The number reported is the percent gloss retention after marring, i.e., $100\% \times$ marred gloss/original gloss.

Acid resistance of coated panels was measured using the
 10 following method: Test panels were spotted with one drop each of 0.625N hydrochloric acid, 0.625N sulfuric acid, sulfuric acid solution at pH 2, and a sulfuric acid solution at pH 3. The panels, with drops uncovered, were then placed in electric ovens at 120°F (48.8°C) and one drop of each acid
 15 solution was added to the panels at ten minute intervals. After 30 minutes, the panels were removed from the ovens and were washed with soap and water and dried, then visually rated for degree of acid etch resistance.

Clear coat formulations and composite coating properties
 20 are reported in Table I below. Numbers in parentheses indicate percent solids by weight within the formulation.

TABLE I

<u>Example</u> <u>Ingredient</u>	<u>1</u> <u>Wt.</u>	<u>2</u> <u>Wt.</u>	<u>3</u> <u>Wt.</u>	<u>4</u> <u>Wt.</u>
<u>Component A</u>				
Ethyl 3-ethoxypropionate	53	30	42	48
1-Methoxy-2-propanol acetate	28	22	18	31
TINUVIN 900 ¹	2 (1.0)	2 (1.0)	2 (1.0)	2 (1.0)

¹ 2-(2'-Hydroxy-benzotriazole-2-yl)-4,6-bis(methyl-1-phenylethyl) phenol UV light stabilizer available from Ciba-Geigy Corp.

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TINUVIN 328 ²	5 (2.0)	5 (2.0)	5 (2.0)	5 (2.0)
TINUVIN 292 ³	3.7 (1.5)	3.7 (1.5)	3.7 (1.5)	3.5 (1.5)
MULTIFLOW ⁴	1.0 (0.2)	1.0 (0.2)	1.0 (0.2)	0.9 (0.2)
RK-5345 ⁵	2.1 (0.5)	2.1 (0.5)	2.1 (0.5)	2.2 (0.5)
Acrylic polyol ⁶	205 (56.7)	234 (64.7)	219 (60.6)	208 (60.5)
<u>Component B</u>				
Ethyl 3-ethoxypropionate	12	23	--	23
DESMODUR N-3390 ⁷	51 (18.7)	58 (21.3)	--	52 (19.9)
DESMODUR Z-4370/2 ⁸	87 (24.5)	--	93 (26.2)	44 (13.0)
Tricarbamoyl Triazine ⁹	--	69 (14.0)	64 (13.1)	31 (6.5)

Clear coat properties

Final Viscosity #4 Ford Cup	23.0"	25.1"	24.4"	25.7"
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Composite properties

Mar Resistance	22.8	49.9	43.7	43.6
Acid Resistance	Good	Fair	Fair	Fair

The data in Table I indicates improvements in acid resistance and significant improvements in mar resistance of
5 composite coatings wherein the clear coat is a multi-

² 2-(2'-Hydroxy-3',5'-ditert-amylphenyl) benzotriazole UV light stabilizer available from Ciba-Geigy Corp.

³ Bis(1,2,2,6,6-pentamethyl-4-piperidinyl) sebacate UV light stabilizer available from Ciba-Geigy Corp.

⁴ Copolymer of ethyl acrylate and 2-ethylhexyl acrylate in xylene, available from Monsanto Chemical Company.

⁵ Flow additive available from E. I. DuPont de Nemours and Company, Inc.

⁶ Copolymer comprising 40% hydroxypropyl acrylate, 20% styrene, 19% butyl acrylate, 18.5% butyl methacrylate, 2% acrylic acid, and 0.5% methyl methacrylate, having an equivalent weight of 325.

⁷ Polyisocyanate of 1,6-hexamethylene diisocyanate, available from Bayer USA.

⁸ Polyisocyanate of isophorone diisocyanate, available from Bayer USA.

⁹ Compound having the formula $C_3N_3(NHCOOR)_3$, wherein R is a mixture of methyl and butyl groups.

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component polyisocyanate-polyol composition and contains a tricarbamoyl triazine as in the present invention.

EXAMPLES 5 AND 6

- 5 Examples 5 and 6 illustrate the preparation of various coating compositions in accordance with the present invention which are used in a color-plus-clear composite coating process. Example 5 is a control two-component clear coat containing polyepoxides and polyacid crosslinking agents.
- 10 Example 6 illustrates the preparation of a clear coat containing a tricarbamoyl triazine compound in place of a portion of polyepoxide.

The coating compositions of Examples 5 and 6 were prepared and spray applied as in Examples 1 through 4 above.

- 15 The composite coatings were cured and properties measured as in Examples 1 through 4 above.

Clear coat formulations and composite coating properties are reported in Table II below.

TABLE II

20

Example Material	5 Wt.	6 Wt.
<u>Component A</u>		
Methyl isobutyl ketone	13	13
DOWANOL DPM ¹⁰	25	25
Epoxy functional acrylic resin ¹¹	128 (29.0)	75 (16.9)
TINUVIN 328	7 (2.6)	7 (2.6)
3,4-Epoxy cyclohexyl methyl- 3,4-epoxy cyclohexyl carboxylate	75 (28.8)	43 (16.9)

¹⁰ Dipropylene glycol monomethyl ether, available from Dow Chemical Co.

¹¹ Prepared according to U.S. Patent No. 4,927,868, Example A.

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RK-5345	0.8 (0.33)	0.8 (0.33)
TINUVIN 292	1.1 (0.25)	1.1 (0.25)
MULTIFLOW	1.3 (0.25)	1.3 (0.25)
Tricarbamoyl Triazine	--	98 (19.0)
<u>Component B</u>		
Exxate 600 solvent ¹²	56	56
Methyl isobutyl ketone	42	42
Acid functional polyester ¹³	78 (20.9)	86 (23.3)
1-octene/maleic anhydride copolymer ¹⁴	34 (9.6)	38 (10.7)
Acid functional polyester ¹⁵	27 (8.2)	30 (9.2)
Isostearic acid	9 (3.4)	10 (3.8)
Dodecyldimethyl amine	11 (4.1)	11 (4.1)

Clear coat properties

Final Viscosity #4 Ford Cup	16.0"	15.2"
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Composite properties

Mar Resistance	4.5	19.4
Acid Resistance	Fair	Good

The data in Table II indicates significant improvements in mar resistance and acid resistance of composite coatings wherein the clear coat is a multi-component polyepoxide-polyacid composition and contains a tricarbamoyl triazine, as in the present invention.

EXAMPLES 7 AND 8

Examples 7 and 8 illustrate the preparation of various coating compositions in accordance with the present invention

¹² Acetate ester available from Exxon Chemicals America.

¹³ Prepared according to U.S. Patent No. 5,256,452, Example G.

¹⁴ Prepared according to U.S. Patent No. 4,927,868, Example F.

¹⁵ Prepared according to U.S. Patent No. 4,927,868, Example B.

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which are used as monocoats. Example 7 is a control one-component coating composition containing a polyepoxide and a polyacid crosslinking agent, as well as an aminoplast.

Example 8 illustrates the preparation of a coating
5 composition containing a tricarbamoyl triazine compound in place of the aminoplast.

The coating compositions of Examples 7 and 8 were prepared by mixing the ingredients together in the order listed. The coating compositions were spray applied to
10 electrocoated steel panels at room temperature (67°F, 19.4°C) and at a relative humidity between 30 and 35 percent, to achieve a dry film thickness of 1.6 mil (40.6 microns). Each panel was given a 15 minute flash at room temperature and then cured for 25 minutes at 285°F (140.5°C).
15 The coating properties were measured as in Examples 1 through 4 above.

Formulations and coating properties are reported in Table III below.

TABLE III

20

Example Material	7 Wt.	8 Wt.
BUTYL CELLOSOLVE acetate ¹⁶	7	7
AROMATIC 100 solvent ¹⁷	14	14
Diethylene glycol monobutyl ether	16	16
TINUVIN 328	4 (2.1)	4 (2.1)
Acrylic microgel ¹⁸	11 (2.8)	11 (2.8)

¹⁶ Acetic acid and 2-butoxyethyl ester, available from Union Carbide Chemicals and Plastics Co., Inc.

¹⁷ Blend of aromatic solvents, also called SOLVESSO 100, available from Exxon Chemicals America.

¹⁸ Prepared according to U.S. Patent No. 4,147,688.

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RESIMENE HM-0610 ¹⁹	17 (8.45)	--
Tricarbamoyl Triazine	--	70 (20.1)
Acid functional polyester ²⁰	71 (28.6)	62 (24.7)
Epoxy functional acrylic ²¹	142 (46.7)	122 (40.4)
MULTIFLOW	0.7 (0.2)	0.7 (0.2)
RK-5345	0.6 (0.2)	0.6 (0.2)
Decanedioic acid	0.7 (0.4)	0.7 (0.4)
AEROSIL R812 silica dispersion ²²	6 (1.8)	6 (1.8)
Primary amyl alcohol	8	8
Carbon black dispersion ²³	34	34
70:30 aromatic hydrocarbon:acetic acid (reducer)	11	--
Solvent Naphtha (reducer)	7	--
<u>Coating properties</u>		
Final Viscosity #4 Ford Cup	26.0"	22.7"
Mar Resistance	30.4	32.2
Acid Resistance	Fair	Good

The data in Table III indicates improvements in acid resistance of one-component polyacid-polyepoxide

¹⁹ Methylated melamine-formaldehyde resin available from Monsanto Chemical Co.

²⁰ Prepared according to U.S. Patent No. 5,256,452, Example F.

²¹ Prepared according to U.S. Patent No. 5,256,452, Example B.

²² Highly dispersed hydrophobic amorphous silicon dioxide available from Degussa Corporation.

²³ A dispersion of carbon in a 1:2 mixture of n-butyl acetate and an acrylic grind vehicle made up of about 29.9% styrene, 19.9% 2-ethyl hexyl acrylate, 19.4% butyl methacrylate, 17.9% methyl methacrylate, 10% hydroxyethyl acrylate, 1.9% methacrylic acid, 0.6% acrylic acid, and 0.4% propyleneimine; ground to a Hegman rating of 8+ and dispersed into a mixture of about 58% diisobutyl ketone and about 42% polyester polyol made up of about 65% trimethyl pentanediol and 35% hexahydrophthalic anhydride to a final composition of 16.1% carbon, 68.8% acrylic grind vehicle, and 15.1% polyester polyol, having a solids content of about 63.9%.

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coatings which contain a tricarbamoyl triazine and are used as monocoats, as in the present invention.

EXAMPLES 9 AND 10

5 Examples 9 and 10 illustrate the preparation of various coating compositions in accordance with the present invention which are used in a color-plus-clear composite coating process. Example 9 is a control one-component clear coat containing a polyisocyanate and an acrylic polyol. Example
10 10 illustrates the preparation of a clear coat containing a tricarbamoyl triazine compound in place of a portion of polyisocyanate.

The coating compositions of Examples 9 and 10 were prepared and spray applied as in Examples 1 through 4 above.

15 The composite coatings were cured for 30 minutes at 285°F (140.5°C) and properties measured as in Examples 1 through 4 above.

Clear coat formulations and composite coating properties are reported in Table IV below.

20

TABLE IV

<u>Example Material</u>	<u>9 Wt.</u>	<u>10 Wt.</u>
N-butyl acetate	100	75
TINUVIN 328	8 (2.65)	8 (2.65)
BYK-301 ²⁴	0.6 (0.1)	0.6 (0.1)
TINUVIN 292	1.0 (0.35)	1.0 (0.35)
Acrylic polyol ²⁵	225 (51.4)	260 (59.4)

²⁴ Anti-mar additive available from BYK-Chemie USA.

²⁵ See footnote 6.

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DESMODUR BL-3175A ²⁶	208 (48.6)	120 (28.0)
METACURE T-120 ²⁷	8 (0.5)	0.8 (0.5)
Tricarbamoyl Triazine	--	75 (12.6)
N-butyl acetate (reducer)	160	134

Clear coat properties

Final Viscosity #4 Ford Cup	21.7"	25.2"
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Composite properties

Mar Resistance	70.1	73.4
Acid Resistance	Good	Good

The data in Table IV indicates an improvement in mar resistance of composite coatings wherein the clear coat is a one-component polyisocyanate-polyol composition and contains 5 a tricarbamoyl triazine, as in the present invention.

EXAMPLES 11 AND 12

- Examples 11 and 12 illustrate the preparation of various coating compositions in accordance with the present invention 10 which are used in a color-plus-clear composite coating process. Example 11 is a control one-component clear coat containing a polyepoxide and polyacid crosslinking agent. Example 12 is the same as Example 11, further containing a tricarbamoyl triazine.
- 15 The coating compositions of Examples 11 and 12 were prepared and spray applied as in Examples 1 through 4 above, over a base coat available from PPG Industries, Inc., as

²⁶ 1,6-hexamethylene diisocyanate capped with methyl ethyl ketoxime, available from Bayer USA.

²⁷ Butyltin mercaptide compound, available from Air Products and Chemicals, Inc.

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NHU9517. Each panel was given a 15 minute flash at room temperature and then cured for 30 minutes at 285°F (140.5°C).

Clear coat formulations and composite coating properties are reported in Table V below.

5

TABLE V

<u>Example Material</u>	<u>11 Wt.</u>	<u>12 Wt.</u>
Ethyl 3-ethoxypropionate	61	61
TINUVIN 900	3 (1.3)	3 (1.3)
TINUVIN 328	3 (1.4)	3 (1.4)
Ethyl ester of maleic anhydride octene copolymer ²⁸	27 (8.6)	27 (8.7)
Epoxy functional acrylic ²⁹	117 (33.2)	119 (33.7)
Epoxy functional acrylic ³⁰	55 (18.4)	56 (18.7)
Acid functional polyester ³¹	103 (33.2)	105 (33.9)
n-Propanol	8	--
Primary amyl alcohol	5	--
R812 Silica dispersion	41 (1.5)	41 (1.5)
MULTIFLOW	2.2 (0.5)	2.2 (0.5)
Decanedioic acid	2.2 (1.0)	2.2 (1.0)
Tricarbamoyl triazine	--	22 (5.0)
Xylene (reducer)	13	33
<u>Clear coat properties</u>		
Final Viscosity #4 Ford Cup	24.9"	23.8"

²⁸ Prepared according to U.S. Patent No. 5,256,452, Example J.

²⁹ Prepared according to U.S. Patent No. 5,256,452, Example A.

³⁰ Prepared according to U.S. Patent No. 5,256,452, Example E.

³¹ Prepared according to U.S. Patent No. 5,256,452, Example F.

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Composite properties

Mar Resistance	14.0	40.7
Acid Resistance	Good	Very Good

The data in Table V indicates significant improvements
in mar resistance and acid resistance of composite coatings
5 wherein the clear coat is a one-component polyepoxide-
polyacid composition and contains a tricarbamoyl triazine, as
in the present invention.

WE CLAIM:

1. A curable composition comprising:

(a) about 30 to 90 percent by weight, based on total
5 weight of resin solids, of a film-forming
composition selected from the group consisting of:

(i) a polyepoxide and a polyacid crosslinking
agent;

(ii) an acrylosilane polymer, an acrylic polyol
10 polymer, and optionally an alkylated
melamine-formaldehyde crosslinking agent; and

(iii) a polyisocyanate and a polymer having at
least one group that is reactive with
isocyanate; and

15 (b) about 10 to 70 percent by weight, based on total
weight of resin solids, of a tricarbamoyl triazine
compound of the formula: $C_3N_3(NHCOXR)_3$, wherein X
is nitrogen, oxygen, sulfur, phosphorus, or
carbon, and R is a lower alkyl group having 1 to
20 12 carbon atoms, or mixtures of lower alkyl
groups.

2. The curable composition of claim 1 wherein the
polyepoxide in the film-forming composition of (i) is an
25 epoxy-containing acrylic polymer.

3. The curable composition of claim 1 wherein the
polyepoxide is present in the film-forming composition of (i)
in amounts of about 10 to 90 percent by weight based on total
30 weight of resin solids in the film-forming composition
of (i).

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4. The curable composition of claim 2 wherein the epoxy-containing acrylic polymer is a copolymer of glycidyl acrylate or methacrylate and at least one other polymerizable ethylenically unsaturated monomer.

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5. The curable composition of claim 1 wherein the polyacid crosslinking agent is present in the film-forming composition of (i) in amounts of about 10 to 90 percent by weight based on total weight of resin solids in the film-forming composition of (i).

6. The curable composition of claim 1 wherein the polyacid crosslinking agent in the film-forming composition of (i) is a carboxylic acid-terminated material having an average of greater than two carboxylic acid groups per molecule.

7. The curable composition of claim 6 wherein the carboxylic acid-terminated material is a carboxylic acid-terminated polyester.

8. The curable composition of claim 1 wherein the film-forming composition of (i) is substantially free of basic esterification catalyst.

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9. The curable composition of claim 1 wherein the film-forming composition is a two-package composition.

10. The curable composition of claim 1 wherein the 30 acrylic silane polymer is present in the film-forming composition of (ii) in amounts of about 10 to 90 percent by

weight based on total weight of resin solids in the film-forming composition of (ii).

11. The curable composition of claim 1 wherein the
5 acrylic polyol polymer is present in the film-forming
composition of (ii) in amounts of about 10 to 90 percent by
weight based on total weight of resin solids in the film-
forming composition of (ii).

10 12. The curable composition of claim 1 wherein the
alkylated melamine-formaldehyde crosslinking agent is present
in the film-forming composition of (ii) in amounts up to 50
percent by weight based on total weight of resin solids in
the film-forming composition of (ii).

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13. The curable composition of claim 1 wherein the
polyisocyanate is present in the film-forming composition of
(iii) in amounts of about 10 to 60 percent by weight based on
total weight of resin solids in the film-forming composition
20 of (iii).

14. The curable composition of claim 1 wherein the
polymer having at least one group that is reactive with
isocyanate is present in the film-forming composition of
25 (iii) in amounts of about 40 to 90 percent by weight based on
total weight of resin solids in the film-forming composition
of (iii).

15. The curable composition of claim 1 wherein the
30 curable composition is a transparent top coat applied over a
pigmented base coat in a color-plus-clear composite coating
on a substrate.

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16. The curable composition of claim 15 wherein the substrate has a primer-surfacer coating to which is applied the base coat.

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17. The curable composition of claim 1 wherein the curable composition is a top coat applied over a primer-surfacer coating on a substrate.

10 18. The curable composition of claim 1 wherein X is oxygen.

19. The curable composition of claim 1 wherein R is a mixture of methyl and butyl groups.

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20. Process for applying a composite coating to a substrate, comprising:

applying a first curable coating selected from the group consisting of: primer surfacer and base coat coatings,

20 applying a second coating over the first coat, where the second curable coating composition has:

(a) about 30 to 90 percent by weight, based on total weight of resin solids, of a film-forming composition selected from the group consisting of:

25 (i) a polyepoxide and a polyacid crosslinking agent;

(ii) an acrylosilane polymer, an acrylic polyol polymer, and optionally an alkylated melamine-formaldehyde crosslinking agent; and

30 (iii) a polyisocyanate and a polymer having at least one group that is reactive with isocyanate; and

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- (b) about 10 to 70 percent by weight, based on total weight of resin solids, of a tricarbamoyl triazine compound of the formula: $C_3N_3(NHCOXR)_3$, wherein X is nitrogen, oxygen, sulfur, phosphorus, or carbon, and R is a lower alkyl group having 1 to 12 carbon atoms, or mixtures of lower alkyl groups; and
- 5 curing the composite coating.

10 21. The process of claim 20, wherein after applying the first coating which is a primer surfacer, the primer surfacer is cured before applying the second curable coating.

15 22. The process of claim 21, wherein applying the second coating is on the cured primer surfacer as a monocoat which optionally may contain pigmentation.

20 23. The process of claim 20, wherein after applying the first coating which is a curable base coat the second curable coating is applied as a wet-on-wet coating and both the base coat and the second coating are conjointly cured.

25 24. Coated article having a substrate with a first cured primer surfacer coating and a second cured coating, where the second cured coating is from the curable coating having (a) about 30 to 90 percent by weight, based on total weight of resin solids, of a film-forming composition selected from the group consisting of:

- 30 (i) a polyepoxide and a polyacid crosslinking agent;
- (ii) an acrylosilane polymer, an acrylic polyol polymer, and optionally an alkylated melamine-formaldehyde crosslinking agent; and

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- (iii) a polyisocyanate and a polymer having at least one group that is reactive with isocyanate; and
(b) about 10 to 70 percent by weight, based on total weight of resin solids, of a tricarbamoyl triazine compound of the
5 formula: $C_3N_3(NHCOXR)_3$, wherein X is nitrogen, oxygen, sulfur, phosphorus, or carbon, and R is a lower alkyl group having 1 to 12 carbon atoms, or mixtures of lower alkyl groups.

25. Coated article having a substrate with a first
10 curable coating which is a base coat having thereon a curable clear coat having:

- (a) about 30 to 90 percent by weight, based on total weight of resin solids, of a film-forming composition selected from the group consisting of:
15 (i) a polyepoxide and a polyacid crosslinking agent;
(ii) an acrylosilane polymer, an acrylic polyol polymer, and optionally an alkylated melamine-formaldehyde crosslinking agent; and
20 (iii) a polyisocyanate and a polymer having at least one group that is reactive with isocyanate; and
(b) about 10 to 70 percent by weight, based on total weight of resin solids, of a tricarbamoyl triazine compound of the formula: $C_3N_3(NHCOXR)_3$, wherein X is nitrogen, oxygen, sulfur, phosphorus, or carbon, and R is a lower alkyl group having 1 to 12 carbon atoms, or mixtures of lower alkyl groups.

INTERNATIONAL SEARCH REPORT

International Application No
PCT/US 96/05971

A. CLASSIFICATION OF SUBJECT MATTER

IPC 6 C08G18/78	C08G18/80	C08G18/58	C08G18/64	C08G18/62
C08G18/42	C08G59/50	C09D175/00		

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 6 C08G C08K

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
Y	EP,A,0 604 922 (AMERICAN CYANAMID CO) 6 July 1994 see claims 1-5,8 see page 3, line 4 - line 37 ---	1-4
Y	FR,A,2 189 479 (BAYER AG) 25 January 1974 see claims 1,2 see page 5, line 13 - line 38 ---	1-4
Y	DE,A,22 47 696 (SYNTHOPOL CHEMIE DR KOCH) 2 May 1974 see claims 1,2 see page 4, paragraph 2 - page 5, paragraph 1 ---	1-4

Further documents are listed in the continuation of box C.

Patent family members are listed in annex.

* Special categories of cited documents :

- *A* document defining the general state of the art which is not considered to be of particular relevance
- *E* earlier document but published on or after the international filing date
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- *T* later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
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INTERNATIONAL SEARCH REPORT

International Application No
PCT/US 96/05971

C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	US,A,4 281 076 (KAMIMURA MASAAKI) 28 July 1981 see claims 1,2,6 see column 5, line 6 - line 21 ---	1-4
A	EP,A,0 045 040 (DU PONT) 3 February 1982 see claims 1-7,11 see page 8, line 35 - page 9, line 33 ---	1
A	US,A,4 939 213 (JACOBS III WILLIAM ET AL) 3 July 1990 cited in the application see claims 1-11 see column 6, line 4 - line 65 ---	1
P,A	EP,A,0 665 252 (KANSAI PAINT CO LTD) 2 August 1995 see claims 1-4,19 -----	1

INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No

PCT/US 96/05971

Patent document cited in search report	Publication date	Patent family member(s)		Publication date
EP-A-0604922	06-07-94	CA-A-	2112415	30-06-94
		JP-A-	6228305	16-08-94
FR-A-2189479	25-01-74	DE-A-	2225958	13-12-73
		BE-A-	800035	26-11-73
		CA-A-	1019878	25-10-77
		GB-A-	1390712	16-04-75
		NL-A-	7307274	29-11-73
		US-A-	3893977	08-07-75
DE-A-2247696	02-05-74	NONE		
US-A-4281076	28-07-81	JP-A-	55027375	27-02-80
		JP-A-	55045758	31-03-80
		CA-A-	1111596	27-10-81
EP-A-0045040	03-02-82	US-A-	4346144	24-08-82
		AU-B-	542072	07-02-85
		AU-B-	7312481	28-01-82
		CA-A-	1159186	20-12-83
		JP-B-	1014945	15-03-89
		JP-C-	1531830	24-11-89
		JP-A-	57051760	26-03-82
US-A-4939213	03-07-90	US-A-	5084541	28-01-92
EP-A-0665252	02-08-95	JP-A-	7216301	15-08-95
		US-A-	5519089	21-05-96